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英文論文名 稱:	Electrophoretic deposition of nickel oxide electrodes for electrochemical capacitors
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## 摘要

Nickel hydroxide ( $\text{Ni}(\text{OH})_2$ ) powder is synthesized by co-precipitation method and is deposited onto the stainless steel (SS) substrate by electrophoretic deposition (EPD) and dip coating, respectively. Electrochemical performances and capacitive properties of the deposited  $\text{Ni}(\text{OH})_2$  electrode after annealing at different temperatures for different lengths of time is investigated by cyclic voltammetry (CV). In addition, nickel oxide/carbon fiber ( $\text{NiO}/\text{VGCF}$ ) and nickel oxide/carbon nanotube ( $\text{NiO}/\text{CNT}$ ) composite electrodes are fabricated by EPD in order to enhance the capacitive behavior of the pure  $\text{NiO}$  electrode.

Surface morphology of the synthesized  $\text{Ni}(\text{OH})_2$  powder is platelet-like shape structure observed by SEM. Annealing temperature influences both the crystal structure and the specific surface area of the synthesized  $\text{Ni}(\text{OH})_2$ . XRD results show that the  $\text{Ni}(\text{OH})_2$  converts into  $\text{NiO}$  after annealing at temperature higher than  $300\text{ }^\circ\text{C}$  due to the removal of water from  $\text{Ni}(\text{OH})_2$ . BET analysis indicates that  $\text{Ni}(\text{OH})_2$  powder after annealing at  $300\text{ }^\circ\text{C}$  has the highest specific-surface area compared with other annealing temperatures.

An electrode after annealing at  $300\text{ }^\circ\text{C}$  has better capacitive behavior due to the conversion of  $\text{Ni}(\text{OH})_2$  to  $\text{NiO}$ . The oxidation and reduction peaks during CV scan appear at  $0.45\text{ V}$  and  $0.35\text{ V}$  versus  $\text{Ag}/\text{AgCl}$  reference electrode, respectively. Specific capacitance of an electrode after annealing at  $300\text{ }^\circ\text{C}$  is found to be about  $140\text{ Fg}^{-1}$  at a scan rate of  $10\text{ mVs}^{-1}$ .

The XRD and SEM results indicate that NiO particles are coated on the surface of carbon nanofiber/nanotube by EPD. The high conductivity of VGCF facilitates the redox reactions of NiO in 0.45 V (oxidation) and in 0.35 V (reduction), and therefore, promotes effectively the nickel oxide in the capacitance characteristics.

Compared with NiO electrode, the specific capacitances of NiO/VGCF and NiO/CNT composite electrodes are increased by  $30 \text{ Fg}^{-1}$  and  $10 \text{ Fg}^{-1}$ , respectively in scan rate of  $10 \text{ mVs}^{-1}$ . The results show that VGCF and CNT can improve the capacitive behavior of the composite electrode effectively.